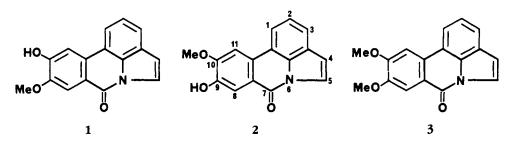
SYNTHESIS OF PRATORIMINE

BALAWANT S. JOSHI, HARIDUTT K. DESAI, and S. WILLIAM PELLETIER*

Institute for Natural Products Research and The Department of Chemistry, The University of Georgia, Athens, Georgia 30602

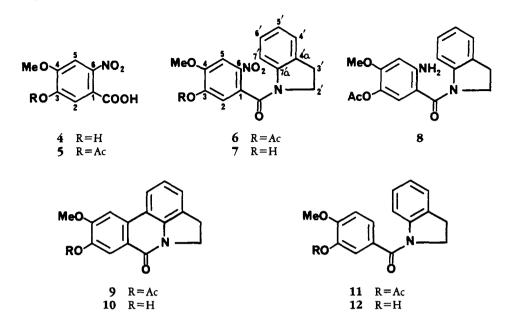
ABSTRACT.—The pyrrolophenanthridone alkaloid pratorimine (2) has been synthesized by an unambiguous route.

Pratorimine and pratorinine were first isolated from *Crinum latifolium* and *Crinum pratense* and assigned structures 1 and 2, respectively, on the basis of ¹H-nmr studies (1,2). Furthermore, the synthesis of pratosine (3) was described, and its partial demethylation with aqueous piperidine to pratorimine was assumed to involve demethylation of the C(10) methoxyl to produce a compound of structure 1. Recently we reported revised structures for pratorimine and pratorinine as 2 and 1, respectively, based on an X-ray diffraction analysis of pratorimine isolated from *Crinum bulbispermum*. The structure of pratorimine was revised on the basis of the identity of the ir spectra of alkaloids isolated from *C. latifolium* and *C. bulbispermum* (3). Ghosal *et al.* (1) appear to have ignored the possibility of demethylation of the C(9) methoxyl group in pratosine (3). We therefore decided to settle this structural ambiguity by a synthesis of alkaloid 2.



RESULTS AND DISCUSSION

Heating methyl 3,4-dimethoxy-6-nitrobenzoate, obtained by methylation and subsequent nitration of veratric acid, with aqueous NaOH gave 3-hydroxy-4-methoxy-6-nitrobenzoic acid (4) (4), which was acetylated to give 5. It has been established that the action of boiling aqueous alkali on 4-nitroveratrole effects exclusively the hydrolysis of the methoxy group in the p-position to the nitro group (5). The acid chloride obtained by treatment of 5 with oxalyl chloride was condensed with indoline to afford 1'-(3-acetoxy-4-methoxy-6-nitrobenzoyl)-2',3'-dihydroindole (6) in good yield. The nitroamide 6 appears to exist as two conformers (¹³C-nmr spectrum) due to restricted rotation. Hydrolysis of 6 gave the corresponding phenol 7, mp 208-209°. Reduction of the nitro group in 6 using Pd-C and hydrogen gave the amine 8. Pschorr cyclization of $\mathbf{8}$ gave the desired phenanthridone $\mathbf{9}$ in poor yield; the major product was probably the deaminated benzoyl indole as in the case of analogous reactions reported earlier (6,7). Hydrolysis of 9 with methanolic KOH gave the phenol 10. In an alternative approach, the amides 11 and 12 were prepared with a view to obtain 10 from the phenol 12. Oxidation of 12 with potassium ferricyanide (8) failed to give the cyclized product. Dehydrogenation of 10 with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone afforded 2 which was shown to be identical with pratorimine.



EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Mp's were taken on a microscope hot-stage apparatus and are corrected. Ir spectra were taken on Perkin-Elmer Model 1420 and nmr spectra were recorded on Varian EM-390 90 MHz, JEOL FT models FX-60 and FX-90 Q spectrometers. Chemical shifts are reported in ppm downfield from internal TMS. Mass spectra were obtained by using a Finnegan Quadrupole 4023 spectrometer.

3-HYDROXY-4-METHOXY-6-NITROBENZOIC ACID (4).—Methyl-3,4-dimethoxy-6-nitrobenzoate (10 g) was added to 10% NaOH (150 ml) and the yellow-orange solution heated under reflux for 24 h. The mixture was poured over crushed ice, acidified with dilute HCl, the precipitate collected and washed with ice H₂O. The crude acid on crystallization from hot H₂O (50 ml) gave 4 (7.5 g), mp 181-182° [lit. (4), mp 181°]; ir (nujol) 3520, 1640, 1580, 1530 cm⁻¹; ¹H nmr (CD₃COCD₃) δ 7.59 (1H, s, H-5), 7.22 (1H, s, H-2), 7.00 (2H, br, OH), 4.02 (3H, s, OCH₃); ¹³C nmr (CD₃COCD₃) 166.7 (COOH), 151.4 (C-4), 149.9 (C-3), 141.8 (C-6), 123.0 (C-1), 116.1 (C-2), 108.5 (C-5), 57.0 (OCH₃).

3-ACETOXY-4-METHOXY-6-NITROBENZOIC ACID (5).—The carboxylic acid 4 (18 g) was heated with Ac₂O (75 ml) and pyridine (15 ml) at 90° for 2 h. The solution was added to crushed ice, the precipitate collected and crystallized from EtOAc to afford 5 (10.7 g), mp 213-214° (dec.); ir (nujol) 1775, 1700, 1610 cm⁻¹; ¹H nmr (CD₃COCD₃) δ 7.65 (1H, s, H-5), 7.59 (1H, s, H-2), 7.35 (1H, br, OH), 4.00 (3H, s, OCH₃), 2.35 (3H, s, COCH₃); ¹³C nmr (CD₃COCD₃+DMSO-d₆) 168.5 (COCH₃), 165.2 (COOH), 155.1 (C-4), 149.3 (C-3), 142.4 (C-6), 125.7 (C-2), 119.4 (C-1), 109.0 (C-5), 57.4 (OCH₃), 20.3 (COCH₃).

1'-(3-ACETOXY-4-METHOXY-6-NITROBENZOYL)-2',3'-DIHYDROINDOLE (**6**).—The carboxylic acid **5** (6 g) in C₆H₆ (60 ml) was heated under reflux with oxalyl chloride (15 ml) for 1 h. The contents were then evaporated under vacuum, and the solidified acid chloride was dissolved in dry Et₂O (150 ml). This solution was added under magnetic stirring to a solution of indoline (5.7 g) in Et₂O (100 ml), during 30 min. The thick precipitate was kept at room temperature for 10 h, collected, and dried (9.8 g). Crystallization from MeOH afforded **6** (7.4 g) as golden yellow plates, mp 176-177°; ir (nujol) 1760, 1640, 1595 cm⁻¹; ¹H nmr (CDCl₃) δ 8.35 (1H, d, J=8 Hz, H-4'), 7.85 (1H, s, H-5), 7.20 (3H, m, Arm. H), 7.25 (1H, s, H-2), 3.96 (3H, s, OCH₃), 3.76 (2H, t, J=7 Hz, N-CH₂), 3.15 (2H, t, J=7 Hz, Ar-CH₂), 2.31 (3H, s, COCH₃). ¹³C nmr (CDCl₃): The major conformer showed: singlets, 167.7 (COCH₃), 163.9 (N-CO), 152.0 (C-4), 144.6 (C-3), 142.6 (C-6), 142.3 (C-4'a), 131.8 (C-7'a), 126.5 (C-1); doublets, 127.7, 124.8, 122.4, 117.3, 113.2, 108.6; triplets, 49.5 (N-CH₂), 28.0 (N-CH₂-CH₂); quartets, 56.7 (OCH₃), 20.5 (COCH₃). The second conformer showed minor signals at: 152.3, 143.3, 140.7, 133.5, 127.2, 125.8, 124.7, 108.8, 48.3, 26.8. Found: C, 60.52, H, 4.57, N, 7.82. C₁₈H₁₆N₂O₆ requires: C, 60.67; H, 4.49; N, 7.86%.

1'-(3-HYDROXY-4-METHOXY-6-NITROBENZOYL)-2',3'-DIHYDROINDOLE (7).—The nitro compound **6** (300 mg) was dissolved in 2.5% methanolic KOH (10 ml) and the deep yellow solution kept at 20° for 2 h. The solution was diluted with H₂O, acidified with dilute HCl, and the precipitate collected. Crystallization from MeOH afforded yellow plates (7; 200 mg), mp 208-209°; ¹H nmr (CD₃COCD₃) δ 8.21 (1H, d, J=8 Hz, H-4'), 7.8 (1H, s, H-5), 7.2 (3H, m, Arm. H), 7.05 (1H, s, H-2), 4.0 (3H, s, OCH₃), 3.8 (2H, t, J=7 Hz, N-CH₂), 3.15 (2H, t, J=7 Hz, Ar-CH₂). ¹³C nmr (CD₃SOCD₃): The major conformer showed: singlets, 164.6 (NCO), 153.5 (C-4), 147.8 (C-3), 142.4 (C-6), 135.7 (C-4'a), 132.3 (C-7'a), 127.9 (C-1); doublets; 127.0, 124.9, 124.0, 116.2, 113.4, 108.2; triplets, 48.9 (N-CH₂), 27.4 (NCH₂-CH₂); quartet, 56.3 (OCH₃). The second conformer showed minor signals at: 140.7, 136.4, 133.3, 125.7, 123.3, 112.5, 47.8, 26.2. Found: N, 8.79; C₁₆H₁₄N₂O₆ requires: N, 8.91%.

1'-(3-ACETOXY-6-AMINO-4-METHOXYBENZOYL)-2',3',-DIHYDROINDOLE (8).—The nitrodihydroindole 6 (3 g) in EtOH (210 ml), and 10% Pd-C (300 mg) was hydrogenated in a Parr apparatus at 1 atmosphere until hydrogen uptake was complete. Usual work up gave on crystallization from $Et_2O/EtOH$ the amine 8 (2.4 g), mp 126-127°; ir (nujol) 3480, 3380, 1760, 1630 cm⁻¹; ¹H nmr (CDCl₃) δ 7.36 (1H, d, J=7 Hz, H-4), 7.00 (3H, m, Arm. H), 6.80 (1H, s, H-2), 6.15 (1H, s, H-5), 4.68 (2H, br, NH₂), 3.90 (2H, t, J=7 Hz, N-CH₂), 3.65 (3H, s, OCH₃), 2.95 (2H, t, J=7 Hz, Ar-CH₂), 2.05 (3H, s, COCH₃); ¹³C nmr (CDCl₃): singlets, 169.5 (C=O), 168.4 (C=O), 153.9 (C-4), 147.1 (C-6), 142.9 (C-7'a), 132.7 (C-4'a), 130.5 (C-3), 110.5 (C-1); doublets, 127.2, 124.9, 123.8, 123.1, 117.1, 100.4; triplets, 50.9 (CH₂-2'), 28.3 (CH₂-3'); quartets, 55.8 (OCH₃), 20.5 (COCH₃). Found: C, 66.18; H, 5.60; C₁₈H₁₈N₂O₄ requires: C, 66.26; H, 5.56%.

9-ACETOXY-10-METHOXY-4, 5-DIHYDRO-7-H-PYRROLO[3,2,1-de]PHENANTHRIDINE-7-ONE (9). —An ice cold solution of NaNO₂ (180 mg in 1 ml H₂O) was gradually added during 15 min to a solution of the amine 8 (200 mg) in glacial HOAc (5 ml) and H₂SO₄ (0.5 ml) at 0°. The yellow solution was heated on the steam bath for 3 h. The tarry reaction mixture was diluted with H₂O, extracted with CH₂Cl₂ and the organic layer was washed with ice cold H₂O and dried over anhydrous Na₂SO₄. Evaporation of the solvent gave a brown colored gum (80 mg). Three batches of this reaction were combined and chromatographed on a "Chromatotron" on a silica gel covered rotor (9) and eluted with CH₂Cl₂ and CH₂Cl₂ containing 2-5% MeOH. Fractions collected were monitored by uv light. One of the fractions (tlc, CHCl₃-5% MeOH Rf 0.3) afforded on crystallization from MeOH, yellowish plates of 9 (25 mg), mp 288-289°; ms, M⁺ m/z 309.

9-HYDROXY-10-METHOXY-4, 5-DIHYDRO-7-H-PYRROLO[3, 2, 1-de]PHENANTHRIDINE-7-ONE (10). —The phenanthridine 9 (20 mg) was suspended in MeOH (2 ml) and was heated at 50° for 3 min with 5% methanolic KOH (0.6 ml). The solution was diluted with ice and acidified with 2 N HCl. The MeOH was removed under vacuum and the precipitate collected, washed with H₂O, and dried to give 10 (18 mg). Crystallization from EtOH afforded pale yellow plates, mp 283-284°; ms, M⁺ m/z 267.

9-HYDROXY-10-METHOXY-7-H-PYRROLO-[3,2,1-de]PHENANTHRIDINE-7-ONE (2).—The phenol 10 (20 mg) was suspended in dry C_6H_6 (50 ml), and the mixture was heated under reflux with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (400 mg) for 10 h. The mixture was evaporated under vacuum, dissolved in CHCl₃, and chromatographed on a "Chromatotron" using a silica gel covered rotor (9). The plate was eluted with CHCl₃, and the band (bluish orange in long wave length uv) corresponding to pratorimine was collected. Evaporation and crystallization from MeOH gave colorless crystals of 2 (3 mg), mp 261-262°, mixture mp with pratorimine 262-263°; ms, M⁺ m/z 265. The ir spectrum (KBr) was superimposable on that of pratorimine.

1'-(3-ACETOXY-4-METHOXYBENZOYL)-2',3'-DIHYDROINDOLE (**11**).—3-Acetoxy-4-methoxybenzoic acid (1.1 g), C_6H_6 (15 ml), and oxalyl chloride (3 ml) were heated under reflux for 3 h. The contents were evaporated under vacuum, the residue dissolved in Et₂O (50 ml) and added with stirring to a solution of indoline (2 g in 20 ml Et₂O). The precipitate was collected, dried, and crystallized from EtOH to afford colorless needles of **11** (1.2 g), mp 194-195°; ms, M⁺ m/z 311; ir (nujol) 1770, 1645, 1612 cm⁻¹; ¹H nmr (CDCl₃) δ 7.91 (1H, dd, J=9, 2 Hz, H-6), 7.65 (1H, d, J=2 Hz, H-2), 7.40 (4H, m, Arm. H), 7.01 (1H, d, J=9 Hz, H-5), 3.92 (3H, s, OCH₃), 3.85 (2H, t, J=9 Hz, N-CH₂), 3.25 (2H, t, J=9 Hz, Ar-CH₂), 2.29 (3H, s, COCH₃); ¹³C nmr (CDCl₃+CD₃SOCD₃): singlets 168.4 (C=O), 167.0 (COCH₃), 154.8 (C-4), 139.2 (C-7'a), 136.4 (C-3), 135.1 (C-4'a), 123.6 (C-1); doublets 129.3, 129.0, 128.1, 125.8, 124.2, 119.5, 111.7; triplets 45.2 (CH₂-2'), 29.1 (CH₂-3'); quartets 56.0 (OCH₃), 20.4 (COCH₃).

1'-(3-HYDROXY-4-METHOXYBENZOYL]-2',3'-DIHYDROINDOLE (12).—A solution of 11 (100 mg) in 2.5% methanolic KOH (6 ml) was warmed to 50° for 5 min and left at 20° for 2 h. This solution was acidified with 10% HCl, the MeOH evaporated under vacuum, and the precipitate collected, washed, and dried. Crystallization from EtOH afforded 12 (80 mg) as colorless cubes, mp 145°; ms, $M^+ m/z$ 269; ir

Journal of Natural Products

(nujol) 3200 (br), 1620, 1590 cm⁻¹; ¹H nmr (CD₃COCD₃) δ 7.82 (1H, dd, J=8, 2 Hz, H-6), 7.22 (6H, m, Arm. H), 4.13 (2H, t, J=8 Hz, N-CH₂), 3.93 (3H, s, OCH₃), 3.15 (2H, t, J=8 Hz, Ar-CH₂); ¹³C nmr (CD₃COCD₃): singlets 169.0 (C=O), 150.1 (C-4), 147.3 (C-3), 144.3 (C-7'a), 133.5 (C-4'a), 131.2 (C-1); doublets 127.5, 125.6, 124.1, 120.0, 117.6, 115.2, 111.9; triplets 51.4 (CH₂-2'), 29.6 (CH₂-3'); quartet 56.3 (OCH₃).

ACKNOWLEDGMENTS

We wish to thank Professor A.W. Frahm for the ir spectrum of pratorimine.

LITERATURE CITED

- 1. S. Ghosal, K.S. Saini, and A.W. Frahm, Phytochemistry, 22. 2305 (1983).
- 2. S. Ghosal, P.H. Rao, D.K. Jaiswal, Y. Kurnar, and A.W. Frahm, Phytochemistry, 20, 2003 (1981).
- 3. J.A. Maddry, B.S. Joshi, A.A. Ali, G.M. Newton, and S.W. Pelletier, Tetrabedron Lett., 26, 4301 (1985).
- 4. M. Greenwood and R. Robinson, J. Chem. Soc., 1371 (1932).
- 5. D. Cardwell and R. Robinson, J. Chem. Soc., 107, 258 (1915).
- L.G. Humber, H. Kondo, K. Kotera, S. Takagi, K. Takeda, W.I. Taylor, B.R. Thomas, Y. Tsuda, K. Tsukamoto, S. Uyeo, H. Yajima, and N. Yanaihara, J. Chem. Soc., 4622 (1954).
- 7. D.H. Hey and D.G. Turpin, Chem. and Ind., 221 (1954).
- 8. T. Kametani, K. Takahashi, S. Shibuya, and K. Fukumoto, J. Chem. Soc., 1800 (1971).
- 9. H.K. Desai, B.S. Joshi, A.M. Panu, and S.W. Pelletier, J. Chromatogr., 322, 223 (1985).

Received 4 October 1985